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# HOLT PHYSICS ANSWERS THE LAW OF THERMODYNAMICS EPUB

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We know that for an ideal gas, work done  $w$  is given as:  $W_{ideal} = -nRT \ln(V_2/V_1)$  And for a van der Waals Gas, work done is given as: Hence for the expansion of a gas,  $V_2 > V_1$ , which shows that numerically the work. Physics Hold Answer is a question and answer site for active. The Thermodynamics chapter of this Holt McDougal Physics Companion Course helps students learn the essential lessons associated with 22 chapters in Holt McDougal Physics. I got really confused that the second law of thermodynamics states that. Time-saving videos related to Holt physics textbook topics

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